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#### **Key indicators**

Single-crystal X-ray study T = 292 K Mean  $\sigma$ (C–C) = 0.003 Å Disorder in main residue R factor = 0.052 wR factor = 0.145 Data-to-parameter ratio = 13.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# Diethyl *cis*-4,5,10,11-tetrahydro-4,11dioxo-1*H*,3*H*-2-oxa-3a,4a,10a,11a-tetraazabenz[*f*]indeno[2,1,7-*ija*]azulene-11b,11c-dicarboxylate

The title compound,  $C_{20}H_{22}N_4O_7$ , is a glycoluril derivative with five fused rings. A benzene ring is fused to a sevenmembered ring which includes two of the N atoms from separate rings of the glycoluril system, and the other two N atoms are linked to an O atom by two methylene groups. The structure is stabilized by intermolecular  $C-H\cdots O$  hydrogen bonds.

## Comment

The glycoluril skeleton has served as an important building block for the preparation of a wide variety of supramolecular systems, including molecular clips (Rowan *et al.*, 1999; Chakraborty *et al.*, 2002; Yin *et al.*, 2006; Li *et al.*, 2006), the cucurbit[*n*]uril family (Freeman *et al.*, 1981), and molecular capsules (Hof *et al.*, 2002; Rebek, 2005). In addition, Isaacs and co-workers have synthesized many methlene-bridged glycoluril dimers from the title compound (Wu, Fettinger *et al.*, 2002). As a part of our ongoing investigation of glycoluril derivatives (Zhou *et al.*, 2005), we report here the structure of the title compound, (I) (Fig. 1).



The crystal packing is stabilized by inversion-related intermolecular  $C-H\cdots O$  hydrogen bonds, forming dimers. These dimers are linked by additional  $C-H\cdots O$  hydrogen bonds involving the carbonyl O atoms of the glycouril system (Table 1 and Fig. 2).

## Experimental

The title compound was synthesized according to the procedure of Wu, Chakraborty *et al.* (2002), using paraformaldehyde (0.18 g, 6 mmol), 1,2-bis(chloromethyl)benzene (0.18 g, 10 mmol), and glycoluril (5.72 g, 20 mmol) as starting materials. Crystals appropriate for data collection were obtained by slow evaporation of a dichloroethane–methanol (4:1  $\nu/\nu$ ) solution at 293 K.

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#### Crystal data

C20H22N4O7  $M_r = 430.42$ Triclinic,  $P\overline{1}$ a = 8.1503 (7) Å b = 11.1548 (9) Å c = 11.9935 (10) Å  $\alpha = 95.016 (1)^{\circ}$  $\beta = 104.297 (1)^{\circ}$  $\gamma = 111.285 (1)^{\circ}$ 

### Data collection

Bruker SMART 4K CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: none 9092 measured reflections

#### Refinement

Refinement on  $F^2$  $w = 1/[\sigma^2(F_0^2) + (0.0791P)^2]$  $R[F^2 > 2\sigma(F^2)] = 0.052$ wR(F<sup>2</sup>) = 0.145 S = 1.05 $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.24 \text{ e} \text{ Å}^{-3}$ 4167 reflections  $\Delta \rho_{\rm min} = -0.36 \text{ e } \text{\AA}^{-3}$ 321 parameters H-atom parameters constrained

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#### Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots $
$\begin{array}{c} C17 - H17B \cdots O2^{i} \\ C19 - H19B \cdots O1^{ii} \\ C20 - H20A \cdots O2^{ii} \end{array}$	0.97	2.51	3.373 (3)	148
	0.97	2.55	3.457 (2)	156
	0.97	2.47	3.217 (2)	133

 $V = 965.42 (14) \text{ Å}^3$ 

 $D_x = 1.481 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

 $\mu = 0.11 \text{ mm}^{-1}$ 

T = 292 (2) K

 $R_{\rm int} = 0.057$ 

 $\theta_{\rm max} = 27.0^{\circ}$ 

+ 0.1548P]

Block colorless

 $0.30 \times 0.20 \times 0.20 \mbox{ mm}$ 

4167 independent reflections

3560 reflections with  $I > 2\sigma(I)$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

Z = 2

Symmetry codes: (i) x - 1, y, z; (ii) -x + 1, -y + 2, -z + 1.

All H atoms were initially located in a difference Fourier map. The methyl H atoms were then constrained to an ideal geometry, with C-H distances of 0.96 Å and  $U_{iso}(H) = 1.5U_{eq}(C)$ , and methyl groups were allowed to rotate freely about their C-C bonds. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C-H distances in the range 0.93–0.97 Å and  $U_{iso}(H) = 1.2U_{eq}(C)$ . The O4 ethoxy group and the O5 atom of the carbonyl group of the other ester are disordered over two orientations. The occupancies of the disordered atoms C13/C13', C14/C14' and O4/O4' refined to 0.650 (11)/0.350 (11), and O5/O5' refined to 0.536 (19)/0.464 (19).

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: PLATON.

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### References

- Bruker (2001). SMART (Version 5.628) and SAINT (Version 6.45). Bruker AXS Inc., Madison, Wisconsin, USA.
- Chakraborty, A., Wu, A., Witt, D., Lagona, J., Fettinger, J. C. & Isaacs, L. (2002). J. Am. Chem. Soc. 124, 8297-8306.



### Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented by circles of arbitrary size. Bonds to atoms of the minor disorder components are drawn as dashed lines.



#### Figure 2

Packing of (I) with hydrogen bonds drawn as dashed lines. For the sake of clarity, H atoms not involved in these interactions are omitted. [Symmetry codes: (i) x - 1, y, z; (ii) -x + 1, -y + 2, -z + 1.] Only one disorder component is shown.

- Freeman, W. A., Mock, W. L. & Shih, N. Y. (1981). J. Am. Chem. Soc. 103, 7367-7368.
- Hof, F., Craig, S. L., Nuckolls, C. & Rebek, J. Jr (2002). Angew. Chem. Int. Ed. 41, 1488-1508.
- Li, Y., Yin, G., Guo, H., Zhou, B. & Wu, A. (2006). Synthesis, pp. 2897-2902. Rebek, J. Jr (2005). Angew. Chem. Int. Ed. 44, 2068-2078.
- Rowan, A. E., Elemans, J. A. A. W. & Nolte, R. J. M. (1999). Acc. Chem. Res. 32, 995-1006
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Wu, A., Chakraborty, A., Witt, D., Lagona, J., Damkaci, F., Ofori, M. A., Chiles, J. K., Fettinger, J. C. & Isaacs, L. (2002). J. Org. Chem. 67, 5817-5830
- Wu, A., Fettinger, J. C. & Isaacs, L. (2002). Tetrahedron, 58, 9769-9777.
- Yin, G., Wang, Z., Chen, Y., Wu, A. & Pan, Y. (2006). Synlett, pp. 49-52.
- Zhou, B.-H., Qu, J.-L. & Wu, A.-X. (2005). Acta Cryst. E61, 03297-03298.